

# Data Report 2018 (draft)

## Chapter 7: Catchment-scale Monitoring

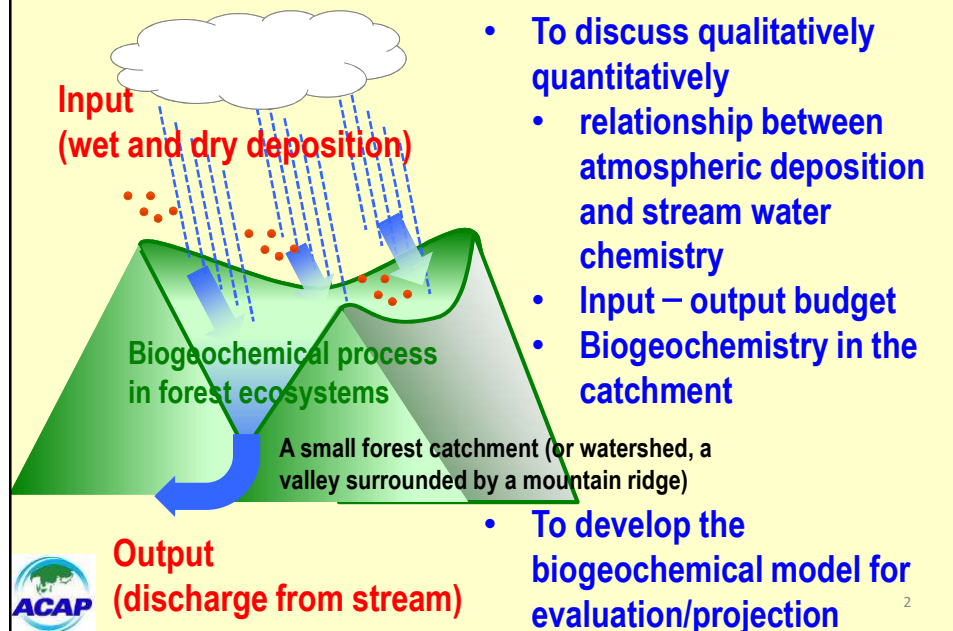
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### Catchment-scale analysis may allow us ...



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### Field operation:

#### Table 7.1 Methods and frequency of major monitoring

Items to be monitored	Recommended methods	Proposed frequency	
<b>Input (total deposition)</b>	Precipitation amount	Rain gauge Continuously or daily	
	Wet deposition <sup>*1</sup>	Wet-only sampling Bulk sampling (when electricity is not available)	Daily or weekly Weekly or biweekly
		Dry deposition <sup>*1</sup>	Inferential method: based on Filter pack data and Meteorological data Automatic monitors (alternative for air concentration)
	<b>Total deposition</b>	Wet + Dry	Biweekly or monthly
Throughfall–Stemflow method (when electricity is not available)		Biweekly	

Note. <sup>\*1</sup>, The relevant EANET data can be referred.




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### Field operation: Table 7.1 (continued)

<b>Output (discharge from the stream)</b>	Water discharge	Weir H-Q curve method (when a weir is not available)	Continuously or biweekly
	Stream water chemistry	Periodical collection of stream water Intensive sampling	Weekly or biweekly During heavy-rain or snow-melting events
	Chemical discharge	Calculation based on the water flux and the chemical concentration	Biweekly, monthly or annual
	<b>Soil</b>	Soil chemistry <sup>*1</sup>	Collection from the permanent plots
<b>Vegetation</b>	Plant growth <sup>*1</sup>	Measurement of tree size (diameter at breast height, height of tree and name of species)	Once for several years

Note. <sup>\*1</sup>, The relevant EANET data can be referred.



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## Laboratory operation:

Table 7.2 Chemical parameters for the respective items

Items	Parameters to be measured
Rainwater samples	<u>Mandatory:</u> $\text{SO}_4^{2-}$ , $\text{NO}_3^-$ , $\text{Cl}^-$ , $\text{NH}_4^+$ , $\text{Ca}^{2+}$ , $\text{Mg}^{2+}$ , $\text{Na}^+$ , $\text{K}^+$ , Electric conductivity (EC), and pH <u>Optional:</u> Total organic nitrogen (TON), Total organic carbon (TOC)
Stream water samples	<u>Mandatory:</u> $\text{SO}_4^{2-}$ , $\text{NO}_3^-$ , $\text{Cl}^-$ , $\text{NH}_4^+$ , $\text{Ca}^{2+}$ , $\text{Mg}^{2+}$ , $\text{Na}^+$ , $\text{K}^+$ , Electric conductivity (EC), pH, and Alkalinity (Gran's ANC and/or pH 4.8 endpoint) <u>Optional:</u> $\text{SiO}_2$ , TON, TOC, Total dissolved Al
Air concentrations (by filter pack method)	Cations: $\text{NH}_4^+$ , $\text{Ca}^{2+}$ , $\text{Mg}^{2+}$ , $\text{Na}^+$ , and $\text{K}^+$ , Anions: $\text{SO}_4^{2-}$ , $\text{NO}_3^-$ , and $\text{Cl}^-$ , Gasses: $\text{SO}_2$ , $\text{HNO}_3$ , $\text{HCl}$ , and $\text{NH}_3$
Soil chemistry	pH ( $\text{H}_2\text{O}$ ), pH (KCl), Exchangeable Base Cations (Ca, Mg, K and Na), Exchangeable Acidity, Al and H, Effective Cation Exchangeable Capacity

*The existing EANET technical documents can also be referred for the laboratory operation!*

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## Data management

- **Units to be used:**
  - Water flux (precipitation amount and stream water discharge): **mm**
  - Chemical concentration:  **$\text{mol}_c \text{L}^{-1}$  ( $\mu\text{mol}_c \text{L}^{-1}$  or  $\text{mmol}_c \text{L}^{-1}$ )**
  - Ion flux (deposition, discharge or soil/vegetation flux):  **$\text{mol}_c \text{ha}^{-1}$  or  $\text{mol}_c \text{m}^{-2}$**
  - Elemental flux for N and S:  **$\text{kg ha}^{-1} \text{y}^{-1}$  or  $\text{g m}^{-2} \text{y}^{-1}$**
- **Water balance on a catchment-scale**
- **Material/elemental balance on a catchment-scale**




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### Monitoring sites:

**Table 7.3 Site for the catchment-scale monitoring**

Country	Site name (and name of river/stream)	Catchment area (km <sup>2</sup> )	Geology and soil type	Coverage of forest (%) and major species	Start year
Japan	Lake Ijira Catchment (Kamagatani River)	2.98	Chert Cambisols (brown forest soils)	99.6%, Japanese cedar ( <i>Cryptomeria japonica</i> ), Japanese cypress ( <i>Chamaecyparis obtusa</i> ), and Japanese red pine ( <i>Pinus densiflora</i> )	November 2007 (the water year here: from November to the next October)



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### Monitoring sites:

**Table 7.4 Survey outline for the catchment-scale monitoring**

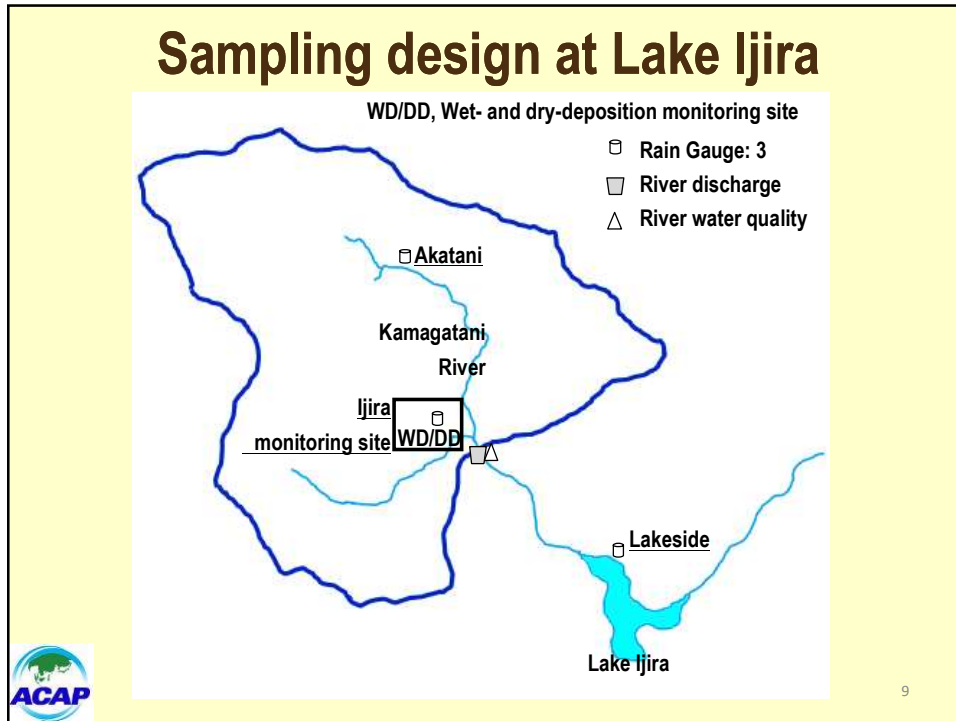
Site	Items	Detailed items
Lake Ijira Catchment	Input (total deposition)	<ol style="list-style-type: none"> <li>1. Precipitation amount (at 3 points)<sup>*1</sup></li> <li>2. Wet deposition<sup>*1</sup></li> <li>3. Dry deposition (Air concentration measurement for Inferential method)<sup>*1</sup></li> <li>4. Total deposition (calculated as wet+dry)</li> </ol>
	Output (chemical discharge from the stream)	<ol style="list-style-type: none"> <li>1. Water discharge (by H-Q curve method)</li> <li>2. Stream water chemistry: biweekly</li> <li>3. Chemical discharge</li> </ol>
	Soil <sup>*2</sup>	Soil chemical properties
	Vegetation <sup>*2</sup>	<ol style="list-style-type: none"> <li>1. Plant growth (field measurement)</li> <li>2. Species composition (field measurement)</li> </ol>

Note: <sup>\*1</sup>, The EANET data collected at the Ijira deposition monitoring site will be used; <sup>\*2</sup>, The EANET data collected in the Lake Ijira soil and vegetation monitoring plots will be referred.



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### Results of monitoring:

**Table 7.5 List of the dataset submitted for Data Report 2017**

Site	Items	Detailed items	Water years
Lake Ijira Catchment	Input (total deposition)	1. Precipitation amount (at 3 points) <i>Calculation of total deposition is under discussion for precise estimation.</i>	2016-2017
	Output (chemical discharge from the stream)	1. Water discharge 2. Stream water chemistry: biweekly 3. Chemical discharge	2016-2017

**Note.**

- The data of wet deposition and dry deposition (air concentration) in the Ijira deposition monitoring site for the period can be found in Data Reports 2016 and 2017
- The water year: from November to October in the following year.
- Soil chemical properties, plant growth and understory-species composition are surveyed in 2016 as a part of soil and vegetation monitoring.

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**Table 7.6 Water balance in the catchment: Lake Ijira Catchment**

Fluxes <sup>*1</sup>	Points	Water year <sup>*2</sup>
		2017-2018
Input by precipitation, mm y <sup>-1</sup>	Akatani (260 m)	3,774
	Ijira monitoring site (140 m)	3,588
	Lakeside (110 m)	3,427
	Mean	3,596
Output by the stream, mm y <sup>-1</sup>	Outlet of Kamagatani River	2,818
Runoff rate, %		78

Note. <sup>\*1</sup>, Fluxes were calculated based on the catchment area, 2.98 km<sup>-2</sup>, at the outlet of Kamagatani River; <sup>\*2</sup>, The water year: from November to the next October.

**Table 7.7: Stream water chemistry** (biweekly data)

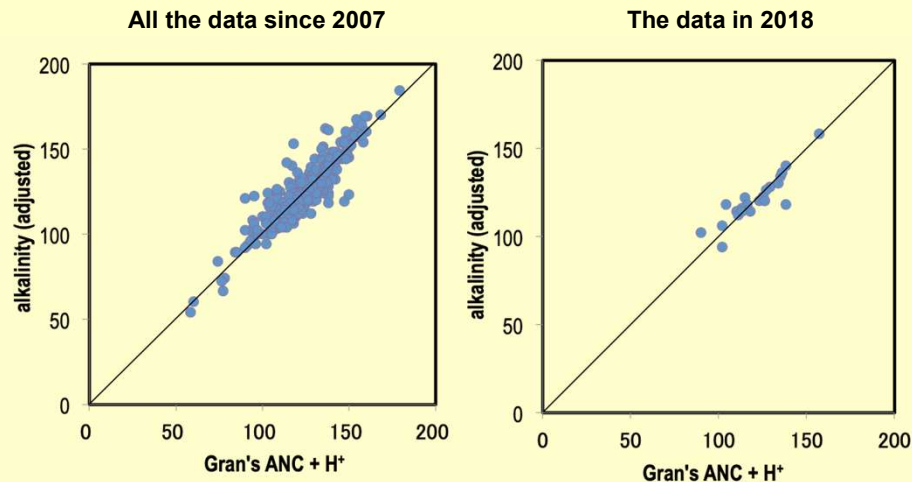
**Table 7.8: Periodic chemical discharges from the stream** (biweekly data)

**Table 7.8: Annual chemical discharges from the stream**

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## Analytical precision of alkalinity at Lake Ijira



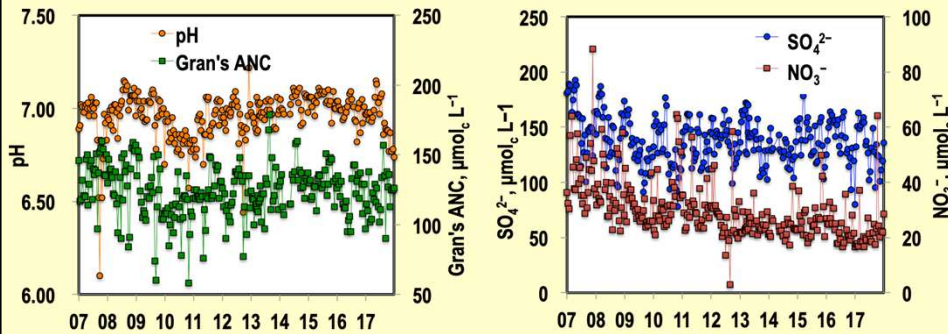
- Analytical precisions of Gran's ANC and pH4.8 alkalinity were fairly good.



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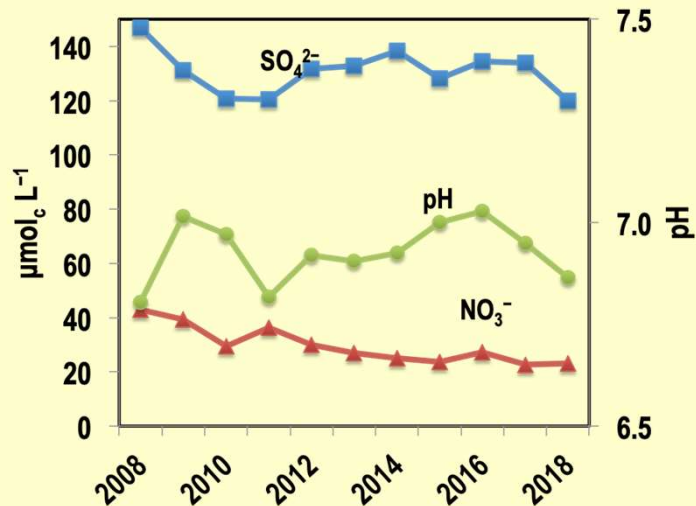
## Recovery from acidification/nitrogen saturation?



- In the mid-1990s, acidification and nitrogen saturation were suggested (Nakahara et al. 2010, *Biogeochemistry*).
- pH and ANC are relatively stable or rather increasing especially for the last several years. Concentrations of  $\text{SO}_4^{2-}$  and  $\text{NO}_3^-$  have been declining.
- Recent recovery is suggested (Sase et al. 2019, *Biogeochemistry*). 13

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## Weighted mean of the stream water chemistry

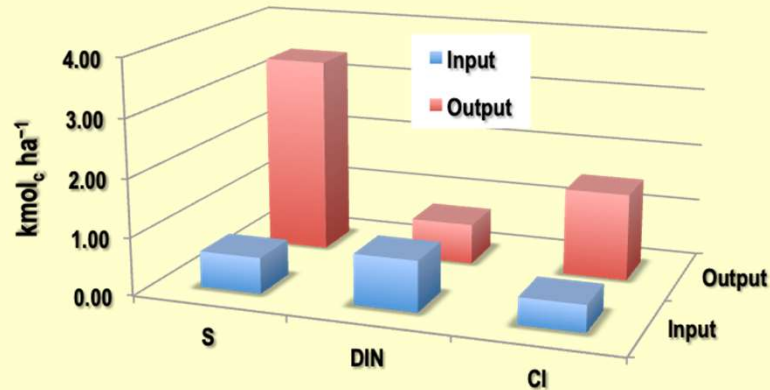


Declining trend of  $\text{NO}_3^-$  concentrations is clear.

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### Comparison with the inputs (wet + dry) WY 2015-2016 (based on Data Report 2016 and 2017)



- S output exceeded the input due to contribution of geology (rock S) (Sase et al. 2019).
- N output is still relatively large. 6.5% of the atmospheric nitrate is exported to the stream directly without processes in the ecosystem (Nakagawa et al. 2018, *Biogeosciences*).

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Thank you for your attention



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